

E.E.E.

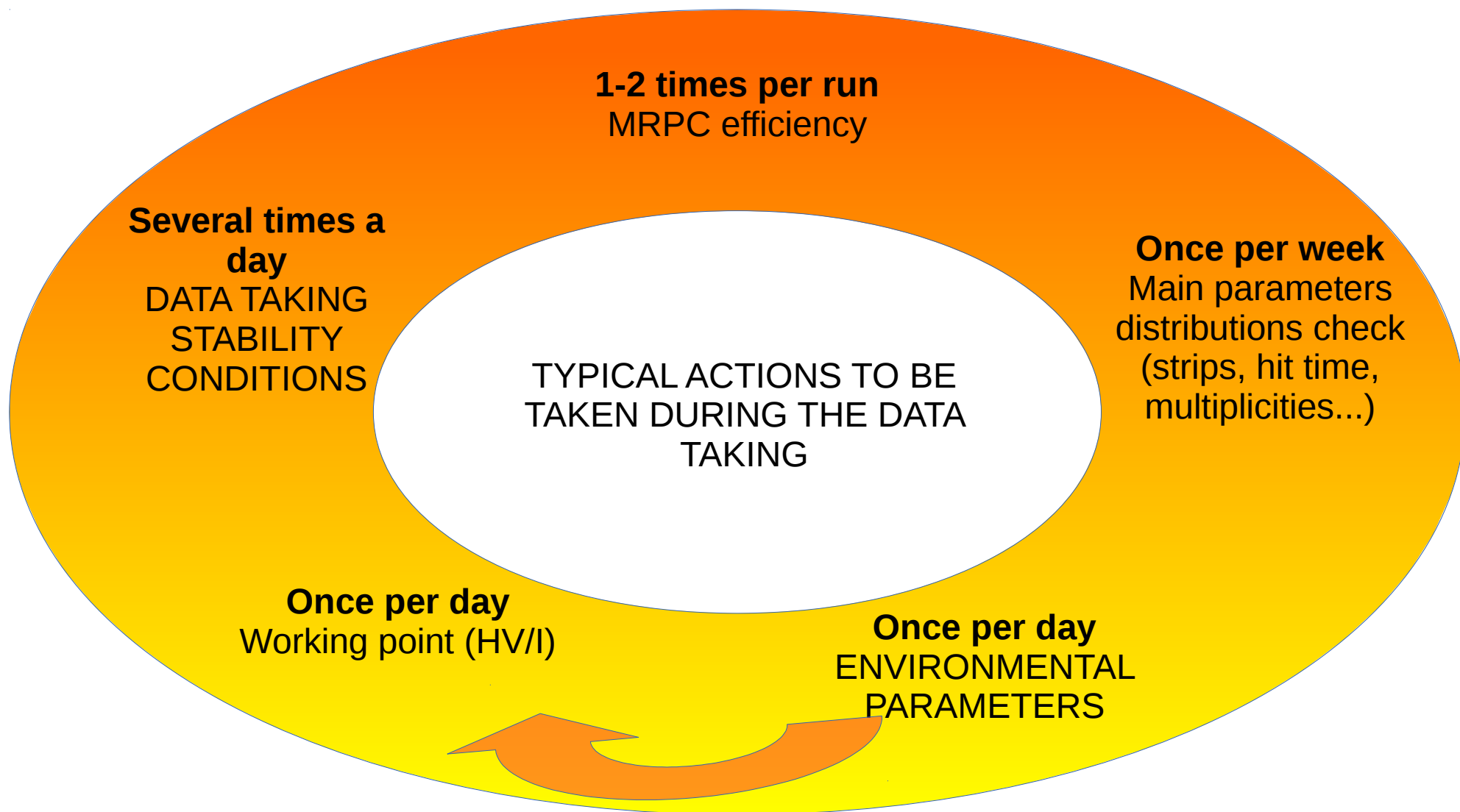
GAS SYSTEM



Data Taking & ShiftList

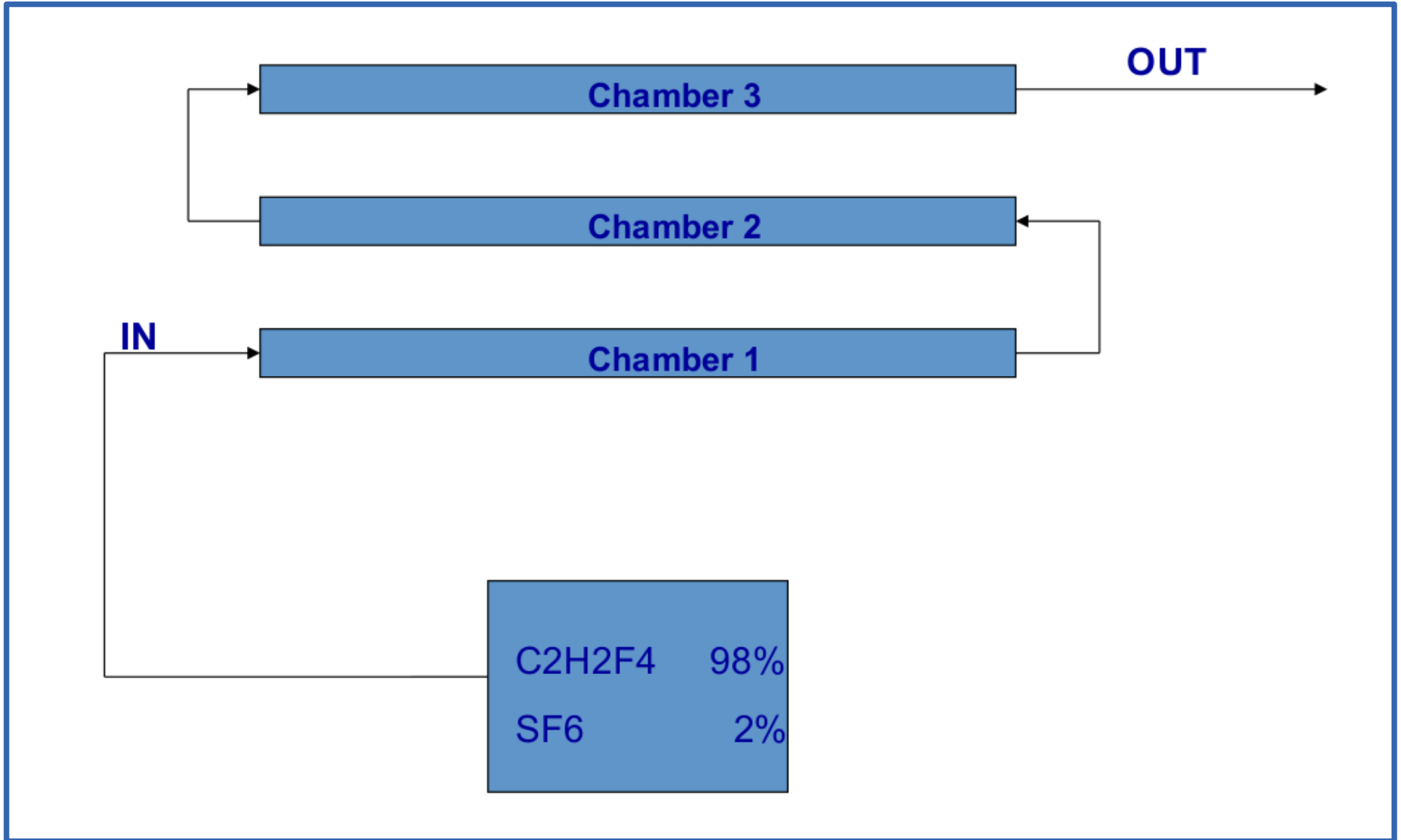
Actions to be taken during the data taking are defined in the so-called

SHIFTLIST



**Data Taking
& ShiftList**

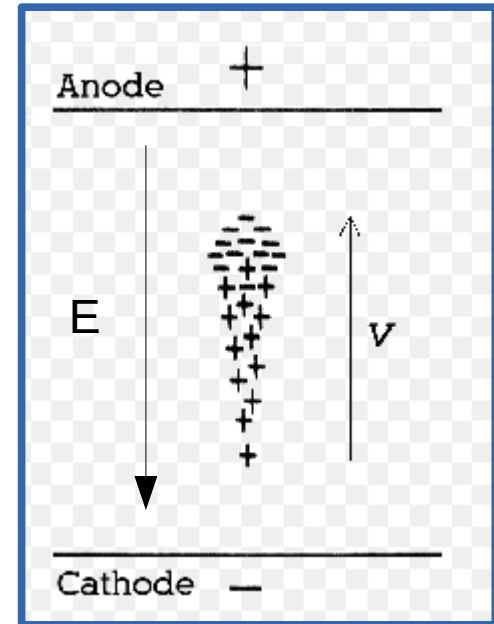
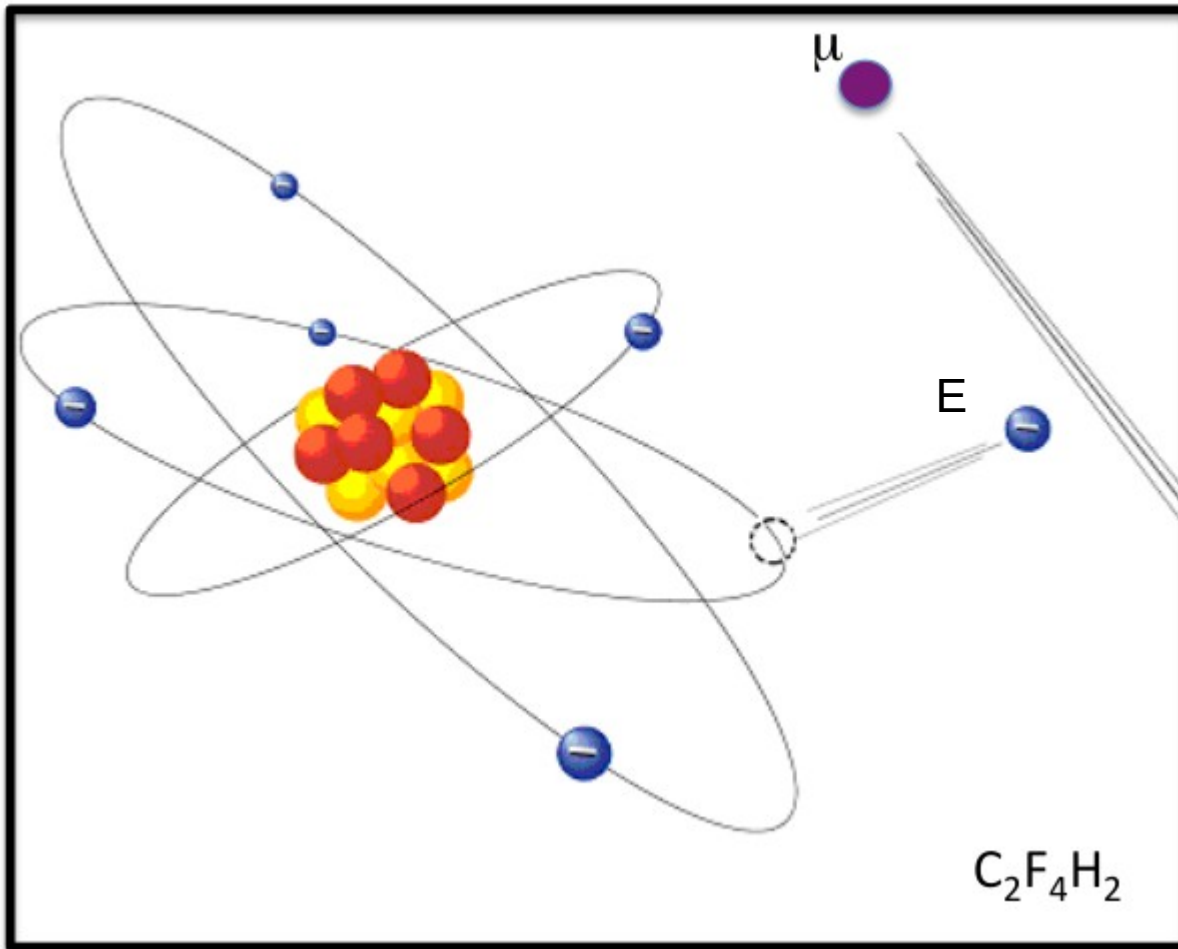
**GAS
SYSTEM**



Data Taking & ShiftList

GAS SYSTEM

WHY SHOULD WE USE A GAS?



The electric field E accelerate electrons (and ions), increasing their kinetic energy

$$qE \times d = qV$$

Which is the energy aquired by electron under 1 kV?

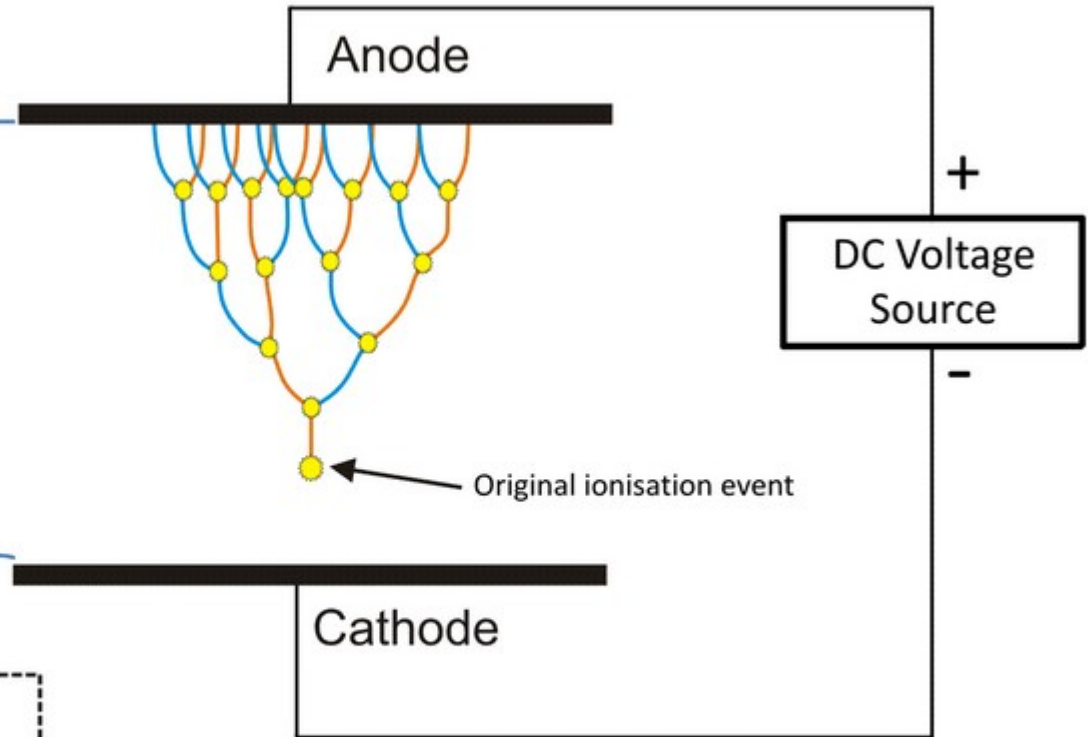
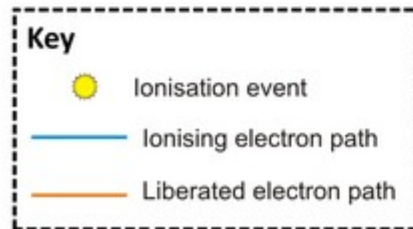
Data Taking & ShiftList

2 GAS SYSTEM

Ionized gas + HV =

Amplifier!

Electric field

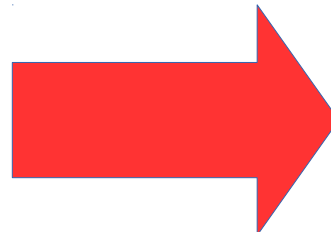


Not to scale

If each electron scattering k secondary electrons are ionized, we have

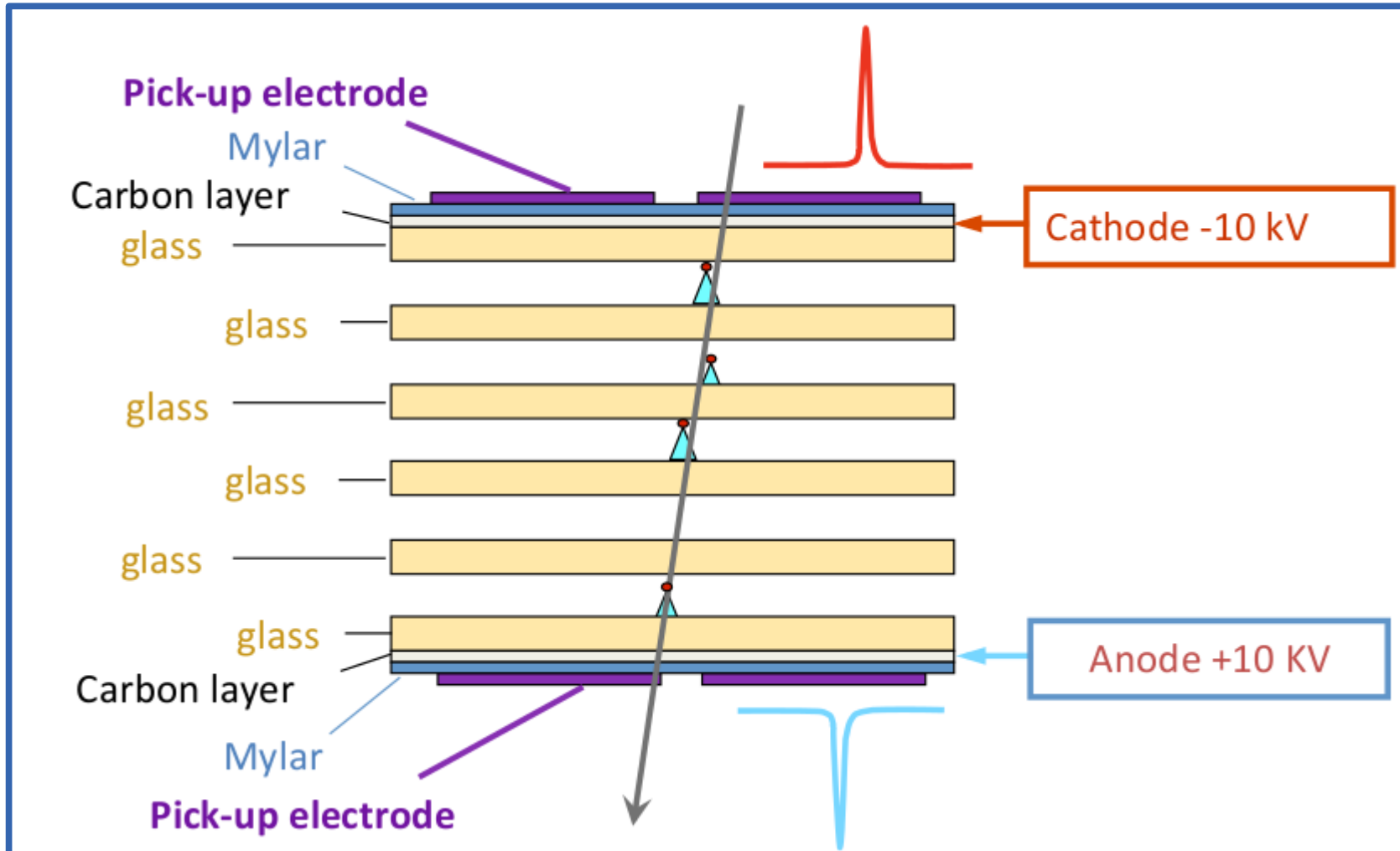
$$N = k^n$$

Electrons after n scatterings



How many scatterings we need if $k=2$ to get amplification = 10^5 ?

Which is the total charge per avalanche?



SF₆ is acting as an **avalanche moderator**, thus shutting the avalanche developing through the gas. This increases the time resolution, decreasing the time and size of the avalanche.

Data Taking & ShiftList

GAS SYSTEM

The gas mixer provides when at 100%

- 6 l/h of freon R134a
- 0.12 l/h of l'SF6

The mix provides

- the avalanche to be started
- and the **quenching** of the avalanche for better time and spatial resolution

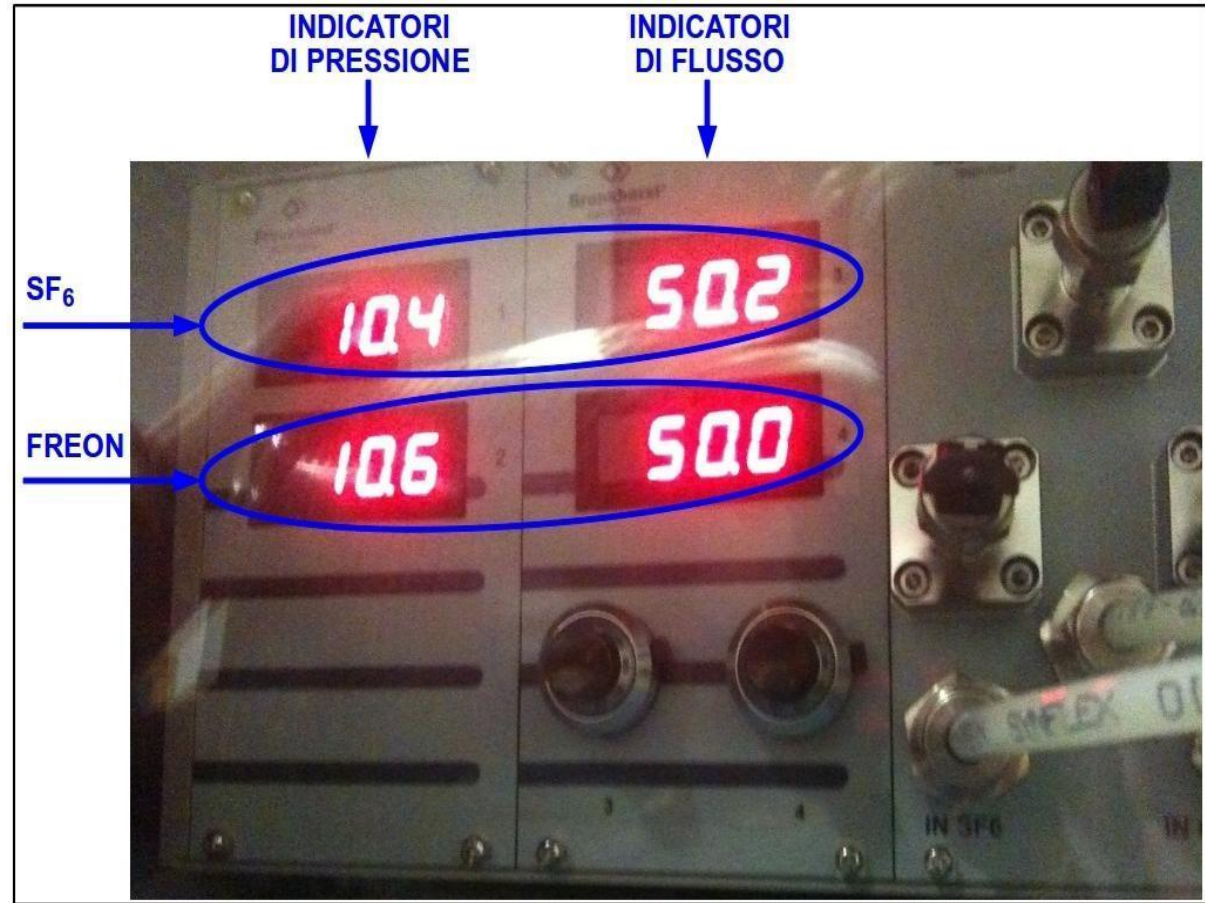
The example shows:

- Input pressure at the mixer in PSI (14.7 PSI = 1 atm)
- Gas mixture flux in percentage with respect to the maxima at 100%

Freon 134a : 50.0% · 6.00 l/h = 3.0 l/h --> 98.03%

SF6 : 50.2% · 0.12 l/h = 0.06024 l/h --> 1.97%

- Gas System parameters to be checked



Reasonable fluxes are

25-30%

Data Taking & ShiftList

GAS SYSTEM

Bottle are equipped by a **pressure reducer**.

Slowly turning the valves the input pressure to the mixer can be adjusted.

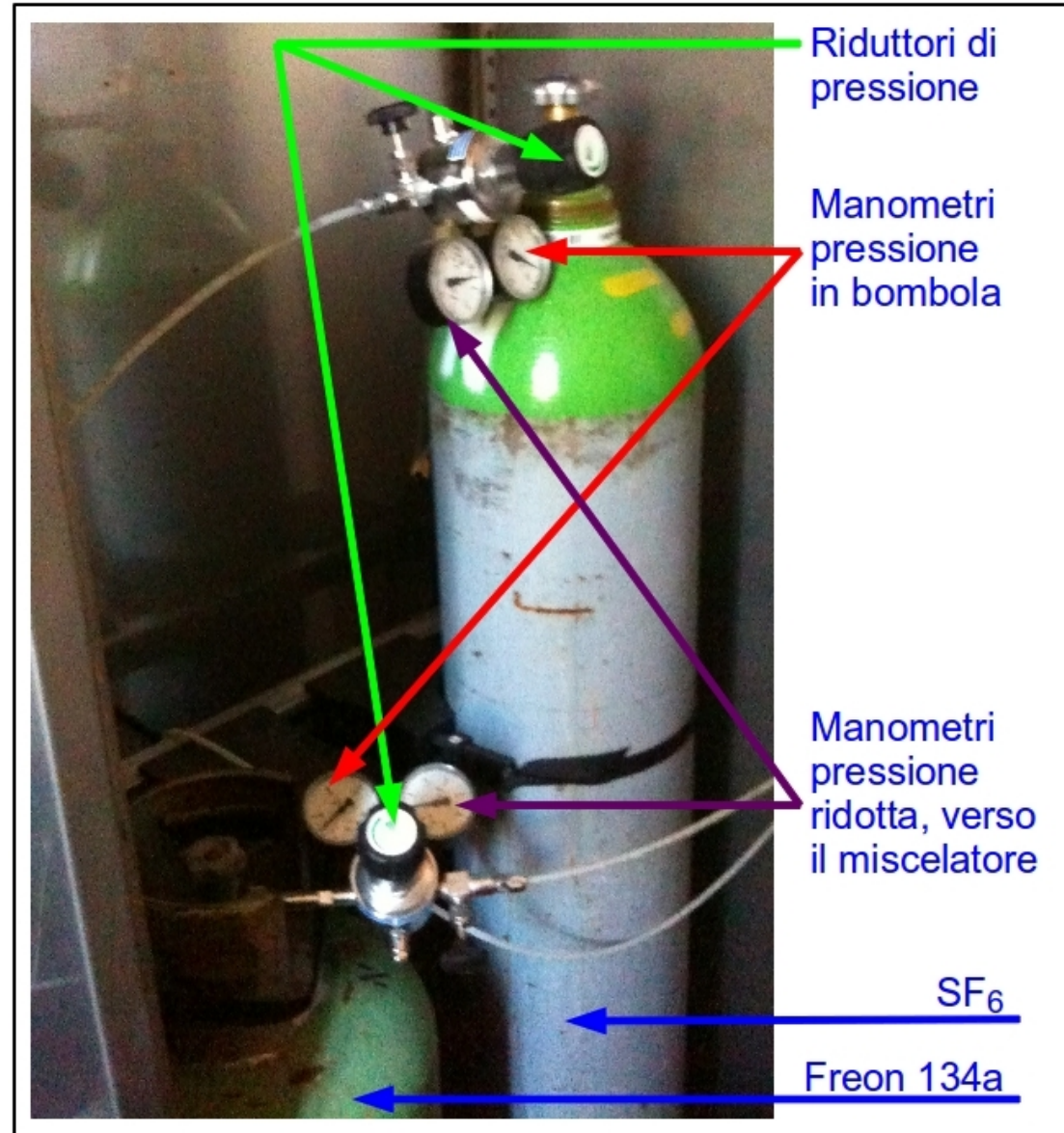
The bottles is exhausted when

- SF6: Internal pressure falls (it takes several years)
- FREON: **net weight of the bottle** allows to forecast the available working days (a bottle lasts around 3-4 months at 30%)

Try to **develop a formula (e.g. in excel sheet)** which gives you the forecast

Hints: use a scale, estimate net weight of gas, daily estimate the gas amount used per day....

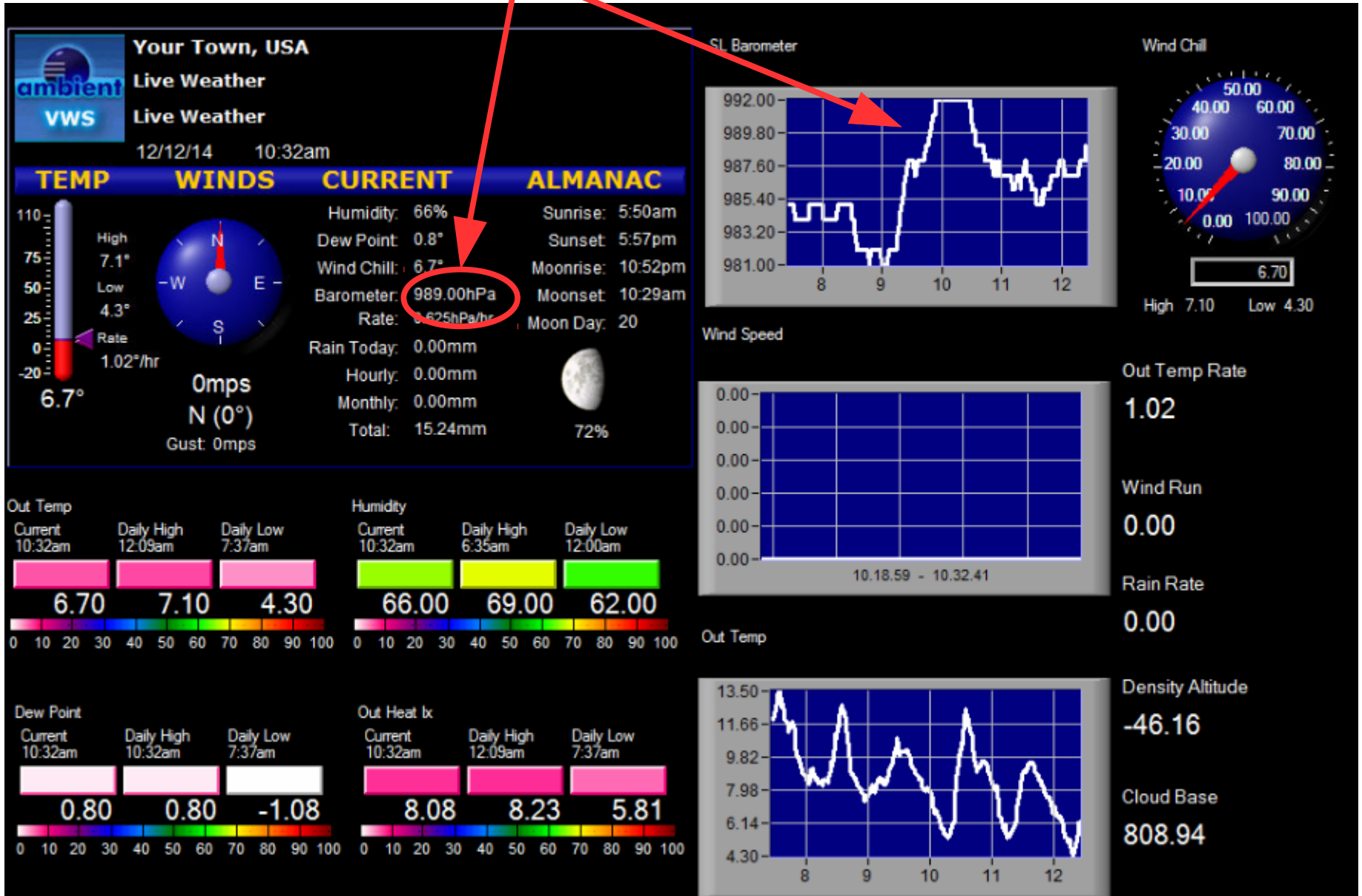
- Bottle pressure check.



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PRESSURE

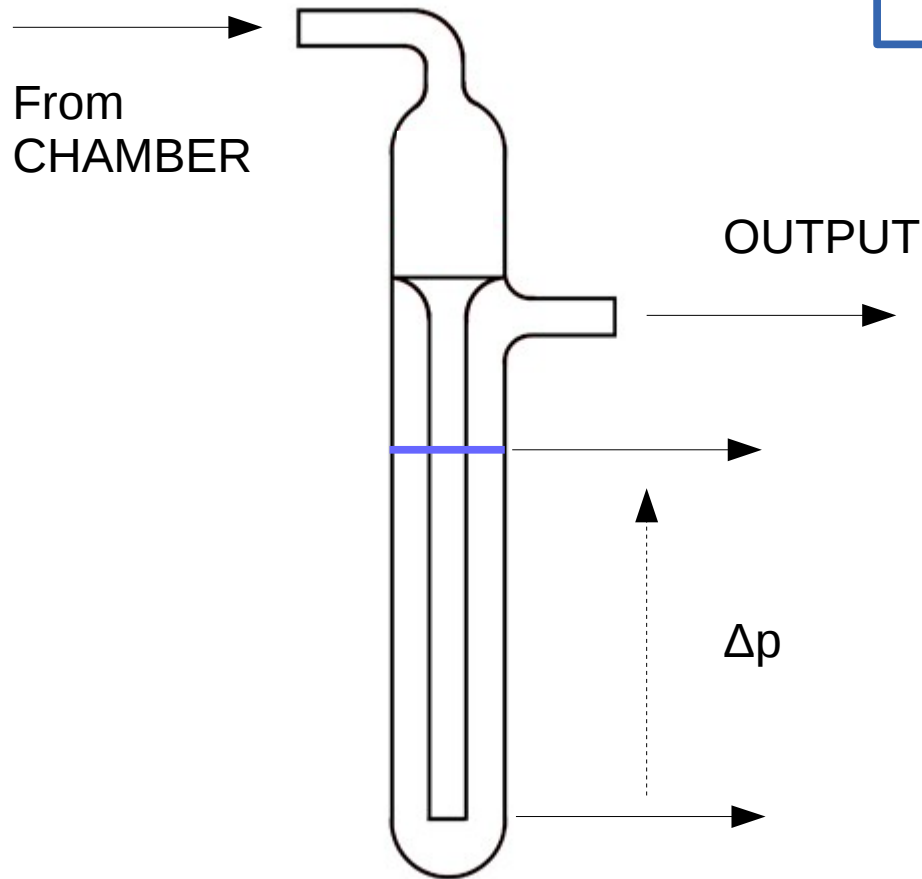
The atmospheric pressure can be a problem for MRPCs, being too high or in case of mixer malfunction can let the air get into the chambers (very rare cases anyhow)



Data Taking & ShiftList

PRESSURE

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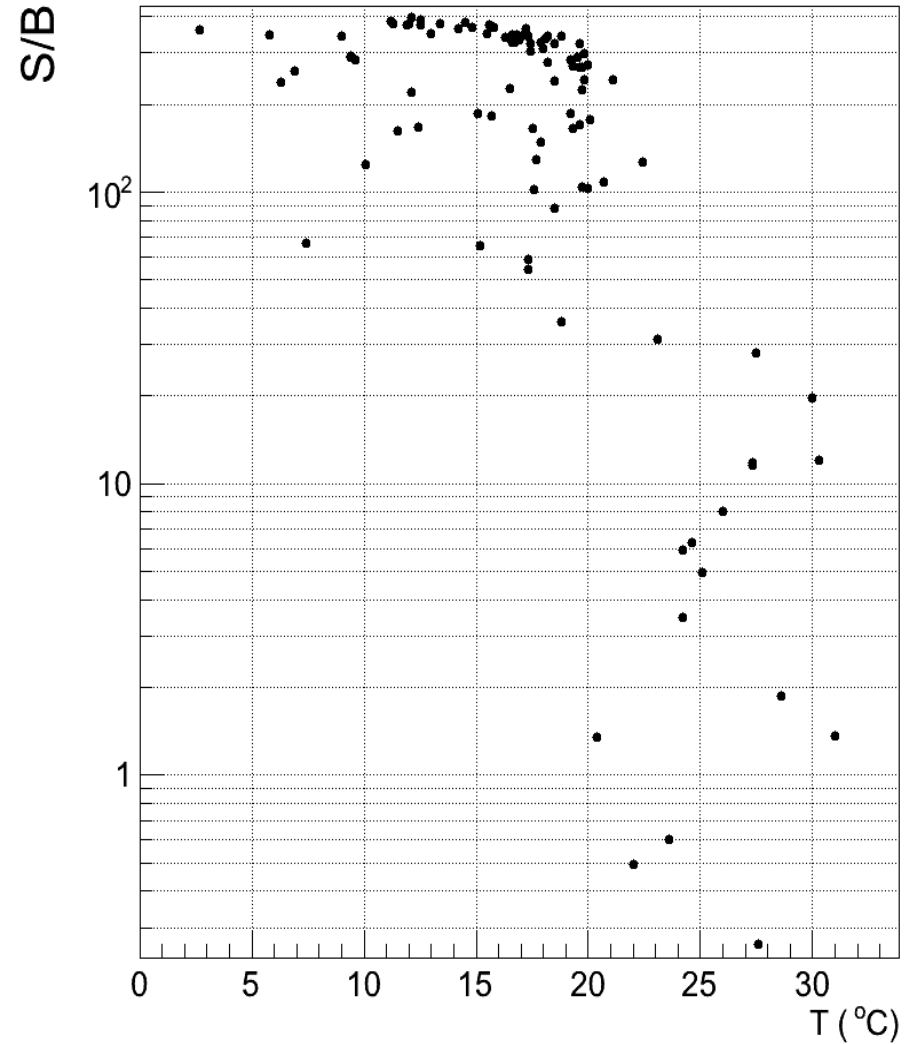
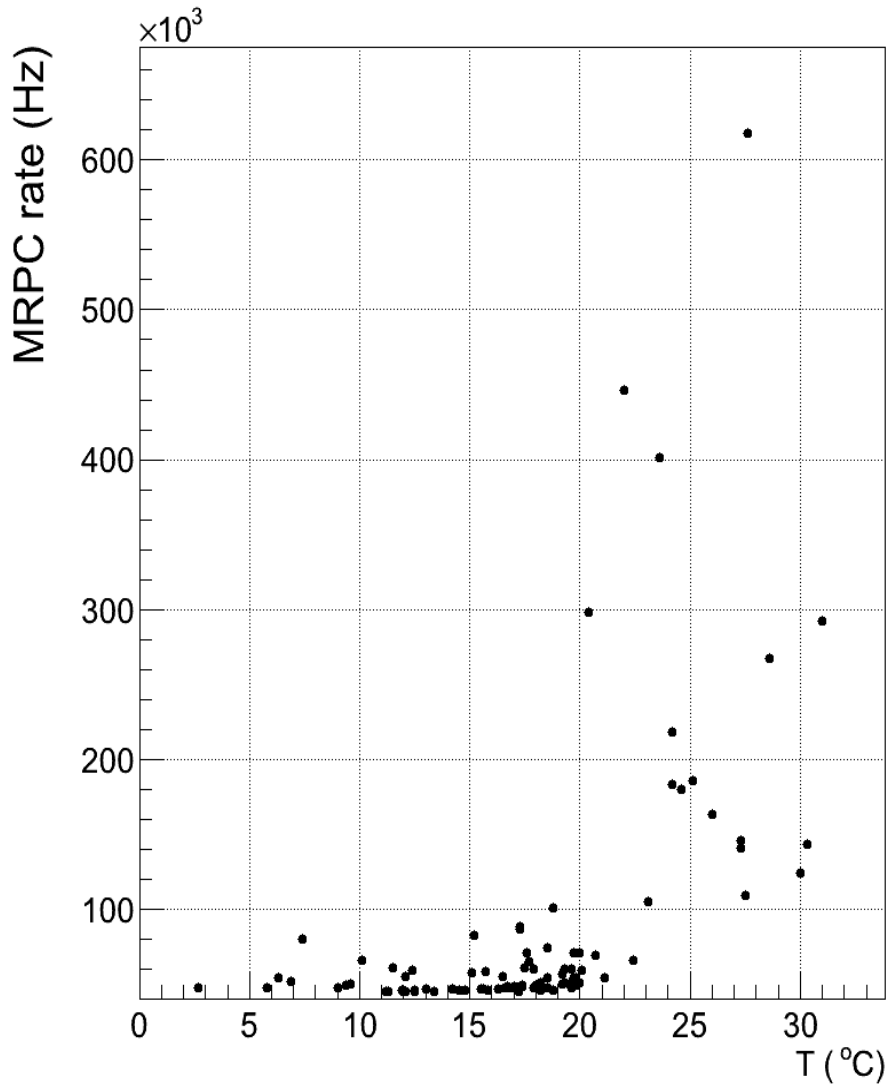
A bubbler can be useful for two reasons:

1. allows to check the gas flow
2. provide a safety pressure gradient among the chamber and the external environment (0.9 mbar/cm in case of paraffin oil)

Data Taking & ShiftList

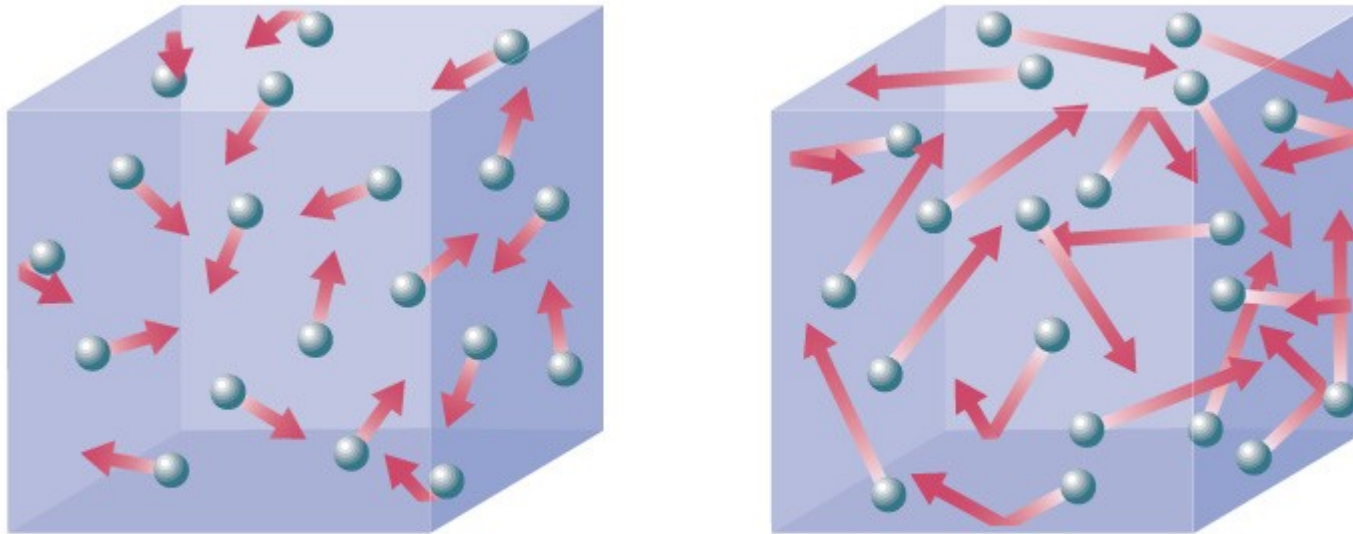
TEMPERATURE

Temperature can be also a problem for
avalanches development



TEMPERATURE

Temperature is a measurement of the kinetic energy of the gas molecules.
Thus high temperature means electrons are easier to be ionized



Longer arrows mean higher average speed.

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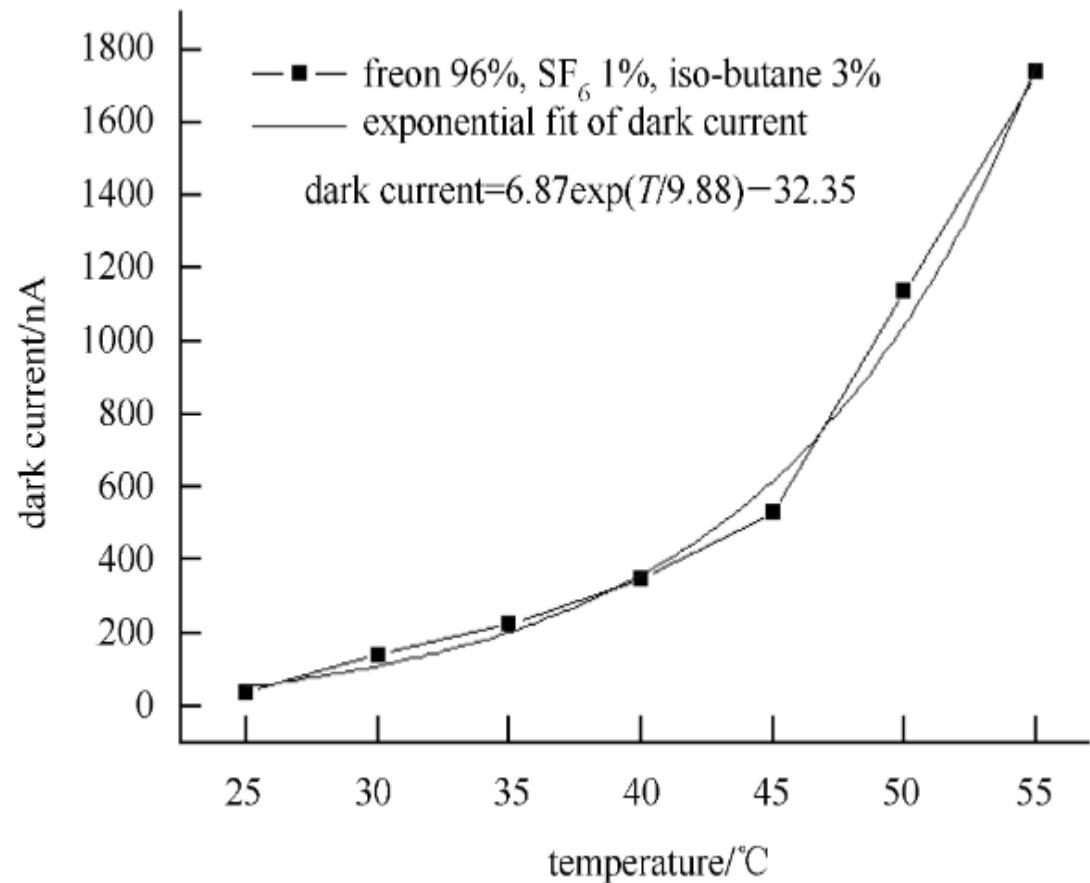
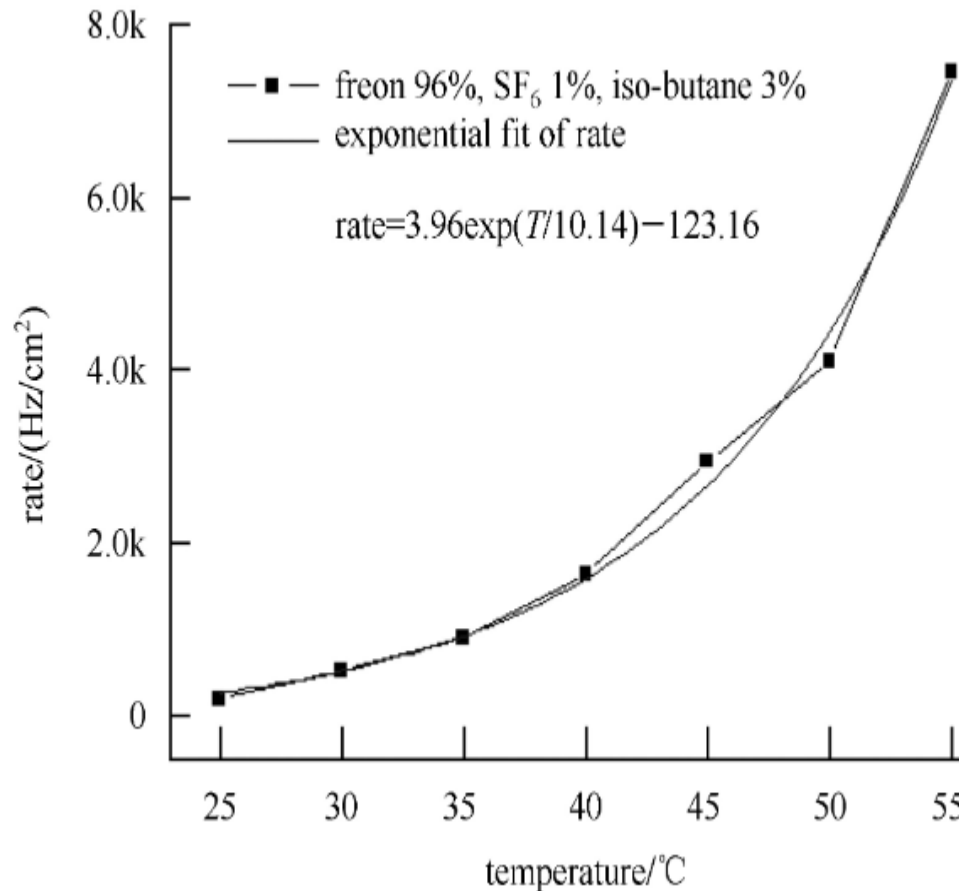
$$\frac{3}{2} kT = \frac{1}{2} mV^2$$

Thus let's try to guess the average speed of Hydrogen atom at:
300 K (env.), 15 MK (sun core), 10^{17} K (LHC)
- the 3rd will give you an incredible result (wrong...) Why?

Data Taking & ShiftList

TEMPERATURE

Thus the temperature of the MRPCs is fundamental to keep the noise within acceptable limits



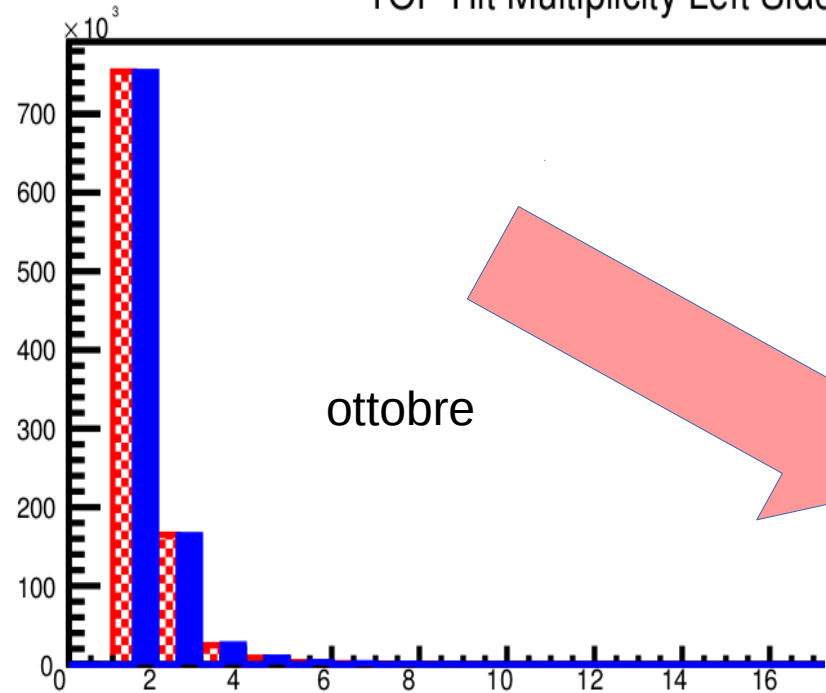
Data Taking
& ShiftList

STRIPS & MULTIPLICITIES

At high temperatures the
multiplicity increases
(number of avalanches per
event)

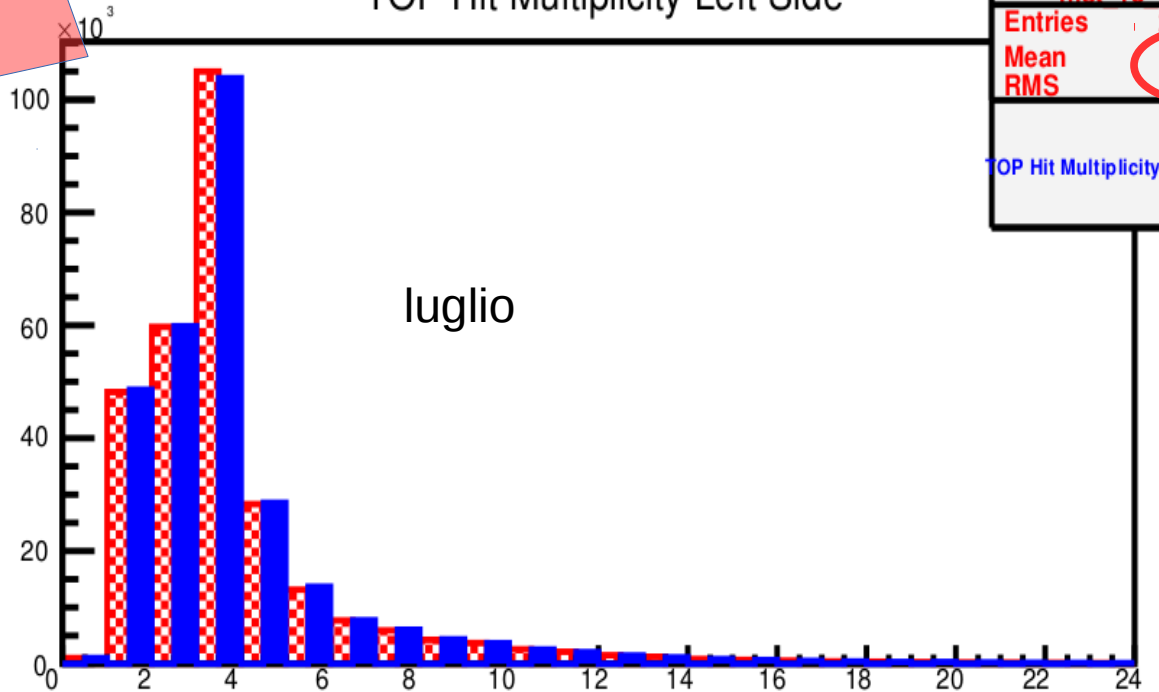
TOP Hit Multiplicity Left Side

hist 13 L	
Entries	959264
Mean	1.295
RMS	0.7788
TOP Hit Multiplicity Right Side	



TOP Hit Multiplicity Left Side

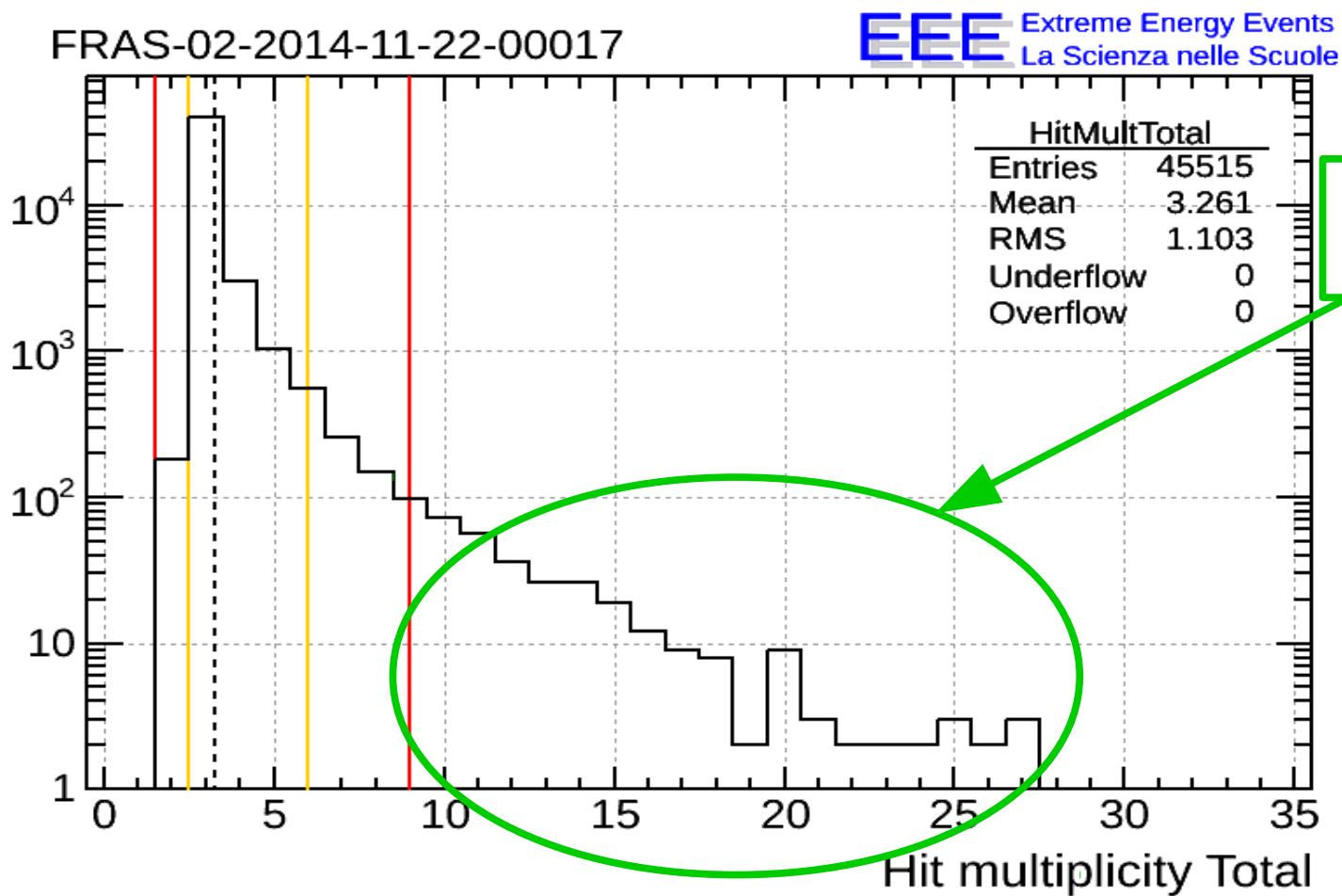
hist 13 L	
Entries	288942
Mean	3.358
RMS	2.657
TOP Hit Multiplicity Right Side	



DQM

RUN DQM

You can check them on run DQM

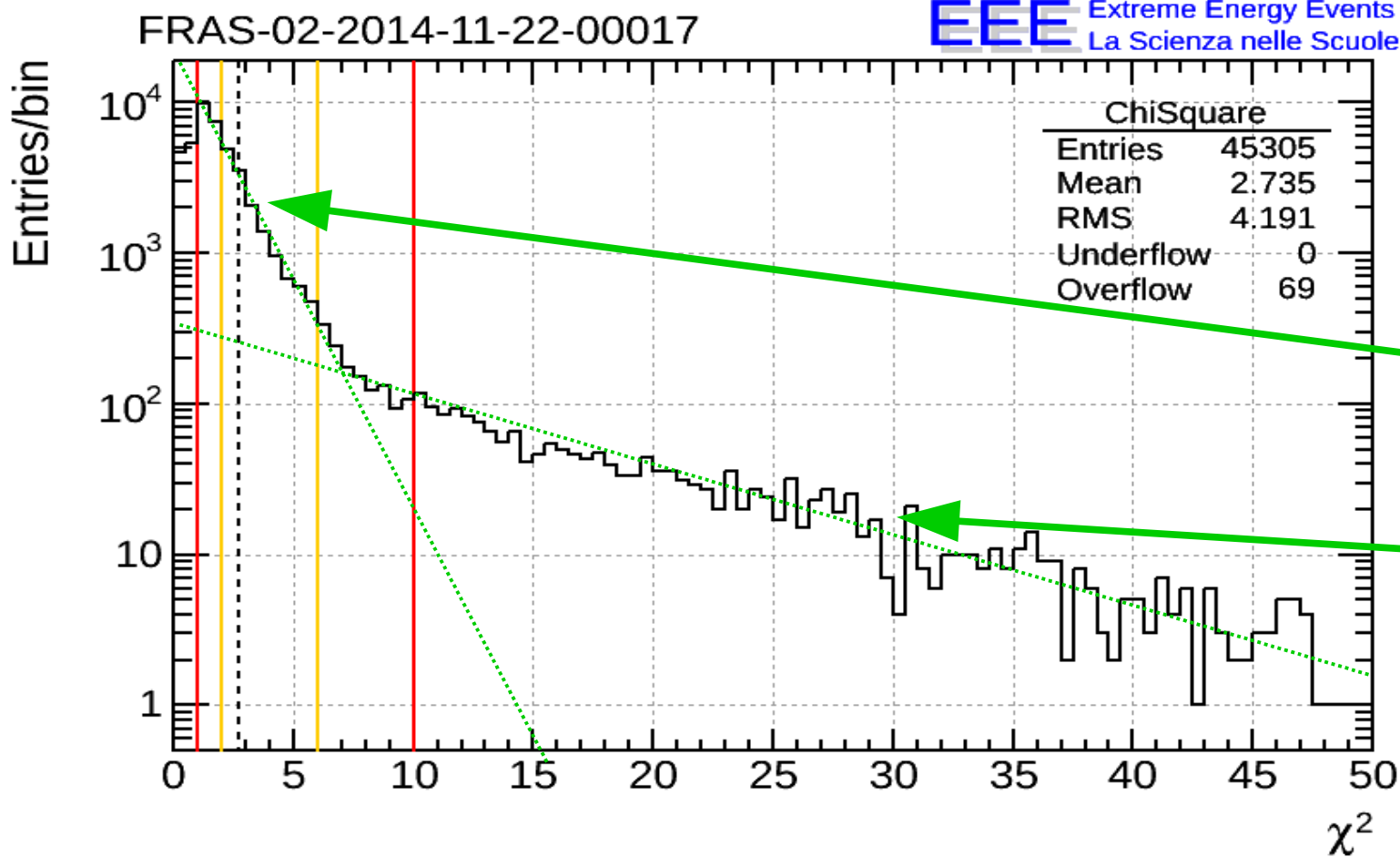
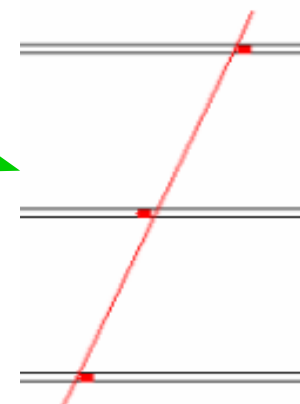


DQM

<http://eee.centrofermi.it/dqm/>

RUN DQM

Also χ^2 distribution of tracks can be effected



THUS.....

TAKE CARE OF GAS SYSTEM

AND

HAVE A GOOD HUNTING FOR NUMBERS!